

**PREPARATION OF *t*-BUTYL 4-DIETHYLPHOSPHONO-3-OXOBUTANETHIOATE AND
USE IN THE SYNTHESIS OF (*E*)-4-ALKENYL-3-OXOESTERS AND MACROLIDES**

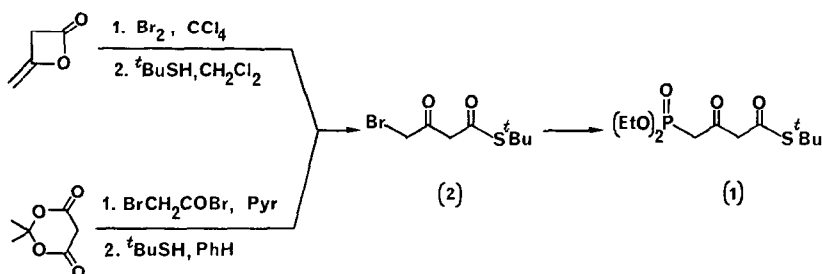
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Abstract: The preparation of *t*-butyl 4-diethylphosphono-3-oxobutanethioate (1) and Wadsworth-Emmons reaction with aldehydes and ketones is reported. Some of the products of these reactions were converted to unsaturated 3-oxomacrolides or 3-oxodiolides by transesterification in the presence of copper(I)-trifluoroacetate.


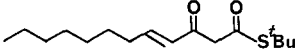
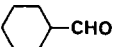
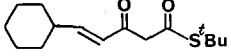
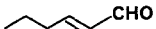
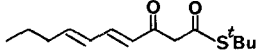
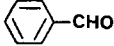
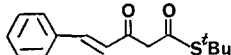

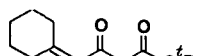
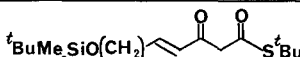

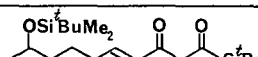
There is a need to develop synthetic units which will permit the ready preparation of 3-oxomacrolides, especially those containing additional double bonds. Several macrolides are known which contain these groupings, or their closely related structures, such as the cephalosporolides,^{1a} diplodialides,^{1b} irumamycin,^{1c} venturicidin^{1d} and kromycin.^{1e} However, to date there is no single synthetic equivalent to these systems. Our interest in *t*-butylacetothioacetate and its use for tetrone acid² and 3-oxodiolide³ synthesis prompted us to develop a related reagent, *t*-butyl 4-diethylphosphono-3-oxobutanethioate (1), the preparation and reactions of which we report here.

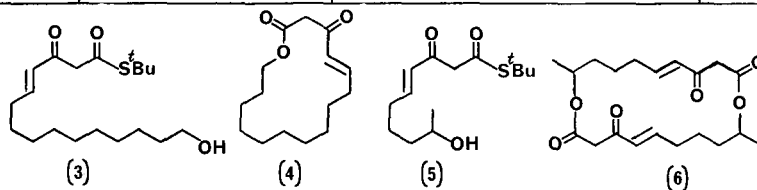
Compound (1) was formed in 85% yield by treatment of the monosodium salt of *t*-butyl 4-bromo-3-oxobutanethioate (2) with sodium diethylphosphite⁴ in THF. The bromothioester (2) was readily available from diketene in 73% yield by bromination⁵ and treatment with *t*-butanethiol. Acylation of Meldrum's acid⁶ with bromoacetyl bromide followed by reaction with *t*-butanethiol also gave (2), in 76% overall yield.



Reaction of (1) with 2 equivalents of sodium hydride in THF at room temperature and Wadsworth-Emmons coupling with various aldehydes and ketones gave moderate to good yields of unsaturated products (Table). In the case of the aldehydes the *E*-geometry of the products was clearly evident from the olefinic coupling constants of 15.5-18 Hz in the ¹H n.m.r. spectra. All products existed as a mixture of keto and enol tautomers.

In order to examine the effectiveness of the *t*-butylthioester group in trans-esterification sequences, particularly for the proposed macrolide synthesis, the product of the reaction of entry 6 in the table was deprotected to the free alcohol (3) in 98% yield using HF in acetonitrile.⁷ This was then treated with copper(I)-trifluoroacetate in buffered (Na₂HPO₄) dichloromethane to give the required unsaturated 3-oxomacrolide (4) (35%). Deprotection of the (*E*)-3-oxothioester from entry 7 using tetra-*n*-butylammonium fluoride⁸ gave (5) (85%) which upon treatment with copper(I)-trifluoroacetate gave the diolide (6) in 35% overall yield.

Entry	Aldehyde/Ketone	Product	Yield%
1			58.3
2			81.0
3			49.5
4			54.4
5			67.2
6	$t\text{BuMe}_2\text{SiO}(\text{CH}_2)_{10}\text{CHO}$	$t\text{BuMe}_2\text{SiO}(\text{CH}_2)_{10}$ 	83.3
7			83.9



While the yields of the macrocyclic products in these reactions are not especially high, other methods were considerably poorer. Analogous chemistry using related phosphorane derivatives such as $\text{Ph}_3\text{P}=\text{CHCOCH}_2\text{COSt}^t\text{Bu}$, failed to afford useful yields of (E)-alkenyl-3-oxo products, like their oxygen counterparts.⁴

We now hope to extend the use of (1) to natural product syntheses.

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References

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